

Evidence for Spontaneous Cycloaromatization of Nine-Membered Monocyclic Enediyne

Takashi Mita, Shinji Kawata, and Masahiro Hiramata*

Department of Chemistry, Graduate School of Science, Tohoku University,
and CREST, Japan Science and Technology Corporation (JST), Sendai 980-8578

(Received June 19, 1998; CL-980464)

Treatment of 2,2-dimethyl-6-nonene-4,8-diynal (**2**) with $\text{CeCl}_3/\text{LiN}(\text{TMS})_2$ at $-5\sim 25^\circ\text{C}$ gave 2,2-dimethyl-1-indanol (**3**) in a moderate yield. In the present study, acyclic (*Z*)-enediyne aldehyde (**2**) was cyclized to afford a nine-membered monocyclic enediyne (**7**), which cycloaromatized spontaneously at room temperature.

Although researches have suggested that a nine-membered monocyclic enediyne (**1**) can undergo Masamune-Bergman cycloaromatization spontaneously at room temperature, no experimental evidence has yet been reported.^{1,2} Construction of the highly strained nine-membered core of bicyclo[7.3.0]dodeca-1,2-diyne system is crucial for synthesizing the enediyne chromophores of extremely potent antitumor chromoprotein antibiotics.^{2b,3} Although several methods for constructing those systems have been reported,⁴ a $\text{CeCl}_3/\text{LiN}(\text{TMS})_2$ mediated intramolecular acetylide addition to aldehyde has been shown to be most effective.^{5,6} All the reaction substrates that lead to cyclononadiyne systems accommodate at least two structural elements which reduce the degree of conformational freedom, i.e., *cis*-olefin or *cis*-epoxide, and an additional five-membered carbocyclic fused ring. We recently demonstrated that the former restriction is not always necessary.⁶ Herein, we report that even acyclic (*Z*)-enediyne aldehyde (**2**) without the latter restriction can be cyclized by $\text{CeCl}_3/\text{LiN}(\text{TMS})_2$, resulting in the formation of an indane derivative (**3**).

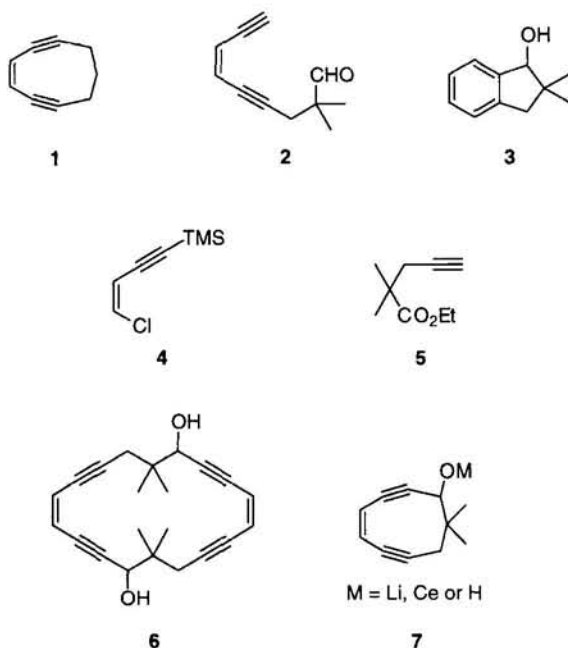
(*Z*)-enediyne (**2**) was synthesized from **4**⁷ and **5**⁸ in a straightforward manner: (i) $\text{Pd}(\text{PPh}_3)_4$, CuI , *n*-BuNH₂,

benzene, r.t., 75%; (ii) $\text{LiOH}\cdot\text{H}_2\text{O}$, THF- H_2O (5:1), r.t., 88%; (iii) DIBAL, Et₂O, -78°C , **2**,⁹ 49%. Treatment of **2** with $\text{CeCl}_3/\text{LiN}(\text{TMS})_2$ (15 equiv.) in THF (5 mmol/l) at -79°C for 1 h yielded only a mixture of dimer (**6**) (27%) and oligomeric products. However, the reaction under a higher dilution (1 mmol/l) and higher temperature ($-5\sim 25^\circ\text{C}$) gave 2,2-dimethyl-1-indanol (**3**)¹⁰ in 14-35% yield. Below -20°C , even under high dilution conditions, primarily **6** and oligomeric products were produced. The best yield of **3** (35%) was obtained by adding **2** to a THF solution of 15 equiv. of $\text{CeCl}_3/\text{LiN}(\text{TMS})_2$ at 25°C over a period of 3 h.¹¹ A large excess of $\text{CeCl}_3/\text{LiN}(\text{TMS})_2$ was necessary to obtain **3** in a yield better than 10%. All attempts to trap the intermediate **7** as a dicobalthexacarbonyl complex using excess $\text{Co}_2(\text{CO})_8$ and NMR detection failed. Only **3** was detected in either case. On the other hand, a trapping experiment of putative indane 4,6-biradical with THF-*d*₈ was attempted but no indane derivative was formed possibly due to a retarded deuterium abstraction.

The present study demonstrates that the intramolecular acetylide-aldehyde condensation reaction with $\text{CeCl}_3/\text{LiN}(\text{TMS})_2$ is applicable to the simple (*Z*)-enediyne aldehyde (**2**) and the resulting nine-membered monocyclic enediyne (**7**) undergoes Masamune-Bergman reaction rapidly at the condensation temperature.

References and Notes

- 1 J. Mayer and F. Sondheimer, *J. Am. Chem. Soc.*, **88**, 603-604 (1966); N. Darby, C. U. Kim, J. A. Salaun, K. W. Shelton, S. Takada, and S. Masamune, *J. Chem. Soc., Chem. Commun.*, **1971**, 1516-1517; R. R. Jones and R. G. Bergman, *J. Am. Chem. Soc.*, **94**, 660-661 (1972); R. G. Bergman, *Acc. Chem. Res.*, **6**, 25-31 (1973).
- 2 a) K. C. Nicolaou, G. Zuccarello, Y. Ogawa, E. J. Schweiger, and T. Kumazawa, *J. Am. Chem. Soc.*, **110**, 4866-4868 (1988). b) K. C. Nicolaou and W.-M. Dai, *Angew. Chem., Int. Ed. Engl.*, **30**, 1387-1416 (1991). c) P. R. Schreiner, *J. Am. Chem. Soc.*, **120**, 4184-4190 (1998) and references cited therein.
- 3 J. W. Grissom, G. U. Gunawardena, D. Klingberg, and D. Huang, *Tetrahedron*, **52**, 6453-6518 (1996).
- 4 a) P. A. Wender, M. Harmata, D. Jeffrey, C. Mukai, and J. Suffert, *Tetrahedron Lett.*, **29**, 909-912 (1988); P. Magnus, and T. Pitterna, *J. Chem. Soc., Chem. Commun.*, **1991**, 541-543; T. Doi and T. Takahashi, *J. Org. Chem.*, **56**, 3465-3467 (1991); K. Mikami, F. Feng, H. Matsueda, A. Yoshida, and D. S. Grierson, *Synlett.*, **1996**, 833-836; A. G. Myers, M. Hammond, Y. Wu, J.-N. Xiang, P. M. Harrington, and E. Y. Kuo, *J. Am. Chem. Soc.*, **118**, 10006-10007 (1996); P. Magnus, R. Carter, M. Davies, J. Elliott, and T. Pitterna, *Tetrahedron*, **52**, 6283-6306 (1996); S. Caddick and V. M. Delisser, *Tetrahedron Lett.*, **38**, 2355-2358 (1997); H. Tanaka, H. Yamada, A. Matsuda, and T. Takahashi, *Synlett.*, **1997**, 381-383. b) For general routes to cyclic enediynes, see: B. Konig, *Angew. Chem., Int. Ed.*



- Engl.*, **35**, 165-166 (1996).
- 5 A. G. Myers, P. M. Harrington, and E. Y. Kuo, *J. Am. Chem. Soc.*, **113**, 694-695 (1991); T. Nishikawa, M. Isobe, and T. Goto, *Synlett.*, **1991**, 393-394.
- 6 K. Iida and M. Hirama, *J. Am. Chem. Soc.*, **116**, 10310-10311 (1994); I. Sato, Y. Akahori, K. Iida, and M. Hirama, *Tetrahedron Lett.*, **37**, 5135-5138 (1996); S. Kawata, F. Yoshimura, J. Irie, H. Ehara, and M. Hirama, *Synlett.*, **1997**, 250-252.
- 7 A. S. Kende and C. A. Smith, *Tetrahedron Lett.*, **29**, 4217-4220 (1988); D. Guillermin and G. Linstrumelle, *Tetrahedron Lett.*, **26**, 3811-3812 (1985).
- 8 P. Magnus, M. J. Slater, and L. M. Principe, *J. Org. Chem.*, **54**, 5148-5153 (1989).
- 9 **2**: pale yellow oil; ^1H NMR (600 MHz, CDCl_3) δ 1.19 (6H, s), 2.55 (2H, br d, $J=2.2$ Hz), 3.29 (1H, dd, $J=2.3$, 0.9 Hz), 5.75 (1H, ddt, $J=10.9$, 2.3, 0.6 Hz), 5.87 (1H, dtd, $J=10.9$, 2.2, 0.9 Hz), 9.57 (1H, s); ^{13}C NMR (150 MHz, CDCl_3) δ 21.27, 27.81, 45.56, 80.30, 80.84, 84.13, 94.42, 118.01, 121.80, 204.55; FT-IR (neat) ν 3288, 2972, 2216, 1729, 1468 cm^{-1} ; HRMS (EI, 70 eV) Calcd for $\text{C}_{11}\text{H}_{12}\text{O}$ 160.0888 (M^+); Found 160.0884 (M^+).
- 10 **3**: colorless solid; mp. 87-90°C; ^1H NMR (600 MHz, CD_2Cl_2) δ 1.00 (3H, s), 1.16 (3H, s), 2.65 (1H, d, $J=15.5$ Hz), 2.76 (1H, d, $J=15.5$ Hz), 4.67 (1H, br s), 7.17-7.22 (3H, m), 7.32-7.34 (1H, m); ^{13}C NMR (150 MHz, CD_2Cl_2) δ 21.88, 27.16, 45.24, 45.48, 84.10, 125.04, 125.74, 127.17, 128.63, 142.78, 145.53; FT-IR (film) ν 3362, 2960, 2930, 2868, 1719, 1611 cm^{-1} ; HRMS (EI, 70 eV) Calcd for $\text{C}_{11}\text{H}_{14}\text{O}$ 162.1045 (M^+); Found 162.1050 (M^+).
- 11 Experimental procedure: To a solution of $(\text{TMS})_2\text{NH}$ (0.32 ml, 1.5 mmol) in THF (8 ml) was added a hexane solution of butyllithium (0.8 ml, 1.3 mmol) at 0 °C. After the solution was stirred for 25 min at the same temperature, the solution was cooled down to -45 °C. To this solution was added a suspension of dry CeCl_3 (299 mg, 1.2 mmol) in THF (8 ml),¹² and then a solution of **2** (12 mg, 0.076 mmol) in THF (60 ml) was added slowly at room temperature over a period of 3 h. The reaction mixture was further stirred for 1 h and quenched with saturated NH_4Cl aqueous solution.
- 12 T. Imamoto, N. Takiyama, K. Nakamura, T. Hatajima, and Y. Kamiya, *J. Am. Chem. Soc.*, **111**, 4392-4398 (1989).